
**Solid biofuels — Determination of the
water soluble chloride, sodium and
potassium content**

*Biocombustibles solides — Détermination de la teneur en chlorure,
sodium et potassium solubles dans l'eau*

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ISO copyright office
Case postale 56 • CH-1211 Geneva 20
Tel. + 41 22 749 01 11
Fax + 41 22 749 09 47
E-mail copyright@iso.org
Web www.iso.org

Published in Switzerland

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 238, *Solid biofuels*.

Introduction

The elements chlorine, sodium and potassium are present in solid biofuels. They can contribute significantly to utilization problems such as corrosion, fouling and slagging in furnaces. Also, they affect the gaseous emissions from the thermal processes.

The chlorine content in solid biofuels is mainly present as water soluble inorganic salts such as sodium and potassium chlorides or other ion-exchangeable forms. Determination of the water soluble chloride content is thus an alternative and simple method to achieve information of the level of chlorine in solid biofuels. However, the content of water soluble chloride is not to be mistaken for the total content of chlorine in the fuels.

In solid biofuels sodium and potassium can be present as both minerals and salts. The salts of these elements are extractable with water and are readily volatile during thermal conversion. By determination of the water soluble content of sodium and potassium, an estimate of the aggressive content of the elements in relation to potential slagging and fouling problems can be achieved. For some biofuels, such as straw, experience has shown that the water soluble content of sodium and potassium corresponds to the total content of the elements. The content of water soluble sodium and potassium is not to be mistaken for the total content of the elements.

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Solid biofuels — Determination of the water soluble chloride, sodium and potassium content

1 Scope

This International Standard describes a method for the determination of the water soluble chloride, sodium and potassium content in solid biofuels by extraction with water in a closed container and their subsequent quantification by different analytical techniques.

2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 9964-1, *Water quality — Determination of sodium and potassium — Part 1: Determination of sodium by atomic absorption spectrometry*

ISO 9964-2, *Water quality — Determination of sodium and potassium — Part 2: Determination of potassium by atomic absorption spectrometry*

ISO 9964-3, *Water quality — Determination of sodium and potassium — Part 3: Determination of sodium and potassium by flame emission spectrometry*

ISO 10304-1, *Water quality — Determination of dissolved anions by liquid chromatography of ions — Part 1: Determination of bromide, chloride, fluoride, nitrate, nitrite, phosphate and sulfate*

ISO 11885, *Water quality — Determination of selected elements by inductively coupled plasma optical emission spectrometry (ICP-OES)*

ISO 16559, *Solid biofuels — Terminology, definitions and descriptions*

ISO 16993, *Solid biofuels — Conversion of analytical results from one basis to another*

ISO 18134-3, *Solid biofuels — Determination of moisture content — Oven dry method — Part 3: Moisture in general analysis sample*

EN 14780, *Solid biofuels — Sample preparation*

Std. Meth. 4500-Cl- D Standard Methods For the Examination Of Water and Wastewater, 18th Edition 1992. 4500-Cl- D. Potentiometric Method

3 Terms and definitions

For the purposes of this document the terms and definitions given in ISO 16559 and the following apply.

3.1

water soluble chloride, sodium and potassium content

amount of the element which can be extracted with water using the extraction procedure specified in this International Standard

4 Principle

The fuel sample is heated with water in a closed container at 120 °C for 1 h. The concentrations of chloride, sodium and potassium in the obtained water extract are determined by one of the following techniques:

- chloride: ion chromatography (IC) or potentiometric titration with silver nitrate;

NOTE When potentiometric titration with silver nitrate is used, any contents of water soluble bromide and iodide will be included in the determination.

- sodium and potassium: flame emission spectroscopy (FES) or flame atomic absorption spectroscopy (FAAS) or inductively coupled plasma optical emission spectroscopy (ICP-OES).

5 Reagents

5.1 Water, containing negligible amounts of chloride, sodium and potassium i.e. amounts that do not contribute significantly to the determinations. Deionised water normally fulfil this requirement.

6 Apparatus

6.1 Heating oven or autoclave, capable of being maintained at a temperature of (120 ± 5) °C.

6.2 Vessel, made of fluoropolymer with a volume of about 100 ml and provided with a tight screw cap. The vessel and the cap shall be capable of withstanding at least 125 °C (232 kPa). If only the water soluble content of chloride is to be determined, an equivalent low-thermal-expansion borosilicate glass vessel can be used.

6.3 Balance, with a resolution of at least 1 mg.

6.4 General laboratory equipment, such as volumetric flasks and measuring cylinders. If sodium and potassium are to be determined, the use of equipment made of glass shall be avoided.

6.5 Membrane filtering apparatus, with membrane filters of mean pore size 0,45 µm.

7 Preparation of the test sample

The test sample is the general analysis sample with a nominal top size of 1 mm or less, prepared in accordance with EN 14780.

If the results are to be calculated other than on an “as determined” basis, the moisture content of the test sample shall be determined concurrently by the method specified in ISO 18134-3, using another portion of the test sample.

8 Procedure

8.1 Extraction

- a) Weigh, in an empty clean vessel (see 6.2), 1,0 g of the analysis sample to the nearest 1 mg.
- b) Add 50,0 ml water, swirl the content and close the vessel tight.
- c) Leave the closed vessel in a heating oven or an autoclave at 120 °C for 60 min.
- d) Take the closed vessel out of the oven or the autoclave and let it cool down to room temperature.

WARNING — Do not attempt to open the vessel before it is cold.

- e) Transfer the content of the vessel to a 100 ml volumetric flask. Wash the inside of the vessel with small portions of water; add the washings to the volumetric flask and make it up to a volume of 100 ml with water.
- f) Filter a portion of the solution [see 8.1 e)] through a membrane filter of pore size 0,45 µm, discarding the first portion of the filtrate. Alternatively the filtering can be carried out using a syringe equipped with a 0,45 µm pore size filter tip.

NOTE If only the water soluble content of chloride is to be determined, filtering may be omitted or a coarse folded filter paper may be used instead of the membrane filter.

8.2 Detection methods

8.2.1 General

Complete the determination by measuring the concentration of the elements in the prepared solution; for chloride by using one of the methods stated in [8.2.2](#) and for sodium and potassium by using one of the methods stated in [8.2.3](#).

8.2.2 Methods for the determination of chloride concentration

For the determination of the chloride concentration one of the following methods shall be used:

- ion-chromatographic determination according to the principles of ISO 10304-1;
- potentiometric titration with silver nitrate according to methods described in Standard Method 4500-Cl- D.

NOTE Equivalent national standards e.g. References [\[1\]](#) [\[2\]](#) or [\[3\]](#) can be used.

Other methods may be used provided that it can be proved that the results obtained are comparable to results obtained by determinations using one of the above stated methods, within the performance characteristics of these methods.

8.2.3 Methods for the determination of sodium and potassium concentration

For the determination of the concentration of sodium and potassium one of the following methods shall be used:

- ICP-OES according to the principles of ISO 11885;
- FAAS according to the principles of ISO 9964-1 and ISO 9964-2;
- FES according to the principles of ISO 9964-3.

For the instrumental technique used, an initial control for eventual interferences shall be performed using a standard addition method and/or a dilution method.

Other methods may be used provided that it can be proved that the obtained results are comparable to results obtained by determinations using one of the above stated methods, within the performance characteristics of these methods.

8.3 Blank test

Carry out a blank test, using the same procedure and methods as described in [8.1](#) and [8.2](#) but omitting the test portion. This assesses both the contents of the elements in the reagents and any contamination from equipment and in the laboratory atmosphere. This shall not be quantitatively significant.

The measured blank value has to be subtracted from the sample value. At high element level the blank should be less than 10 % of the sample content. For low element level (a content below 500 mg/kg in the sample), it is adequate that the contents of the elements in the blank solution are 30 % or less of the contents of the elements in the sample solution.

9 Calculation

The water soluble content of the element in the sample as analysed, w_x , expressed in mg/kg is given by Formula (1):

$$w_x = \frac{(c - c_0) \times V}{m} \quad (1)$$

where

- c is the concentration of the element (chloride, sodium or potassium), in mg/l, in the extract solution [see 8.1 e)];
- c_0 is the concentration of the element (chloride, sodium or potassium), in mg/l, in the extract solution of the blank experiment (see 8.3);
- V is the volume, in ml, of the extract solution;
- m is the mass, in g, of the test portion used.

Report the results as the mean of duplicate determinations.

The results shall be calculated to a dry basis or to an as received basis according to ISO 16993.

10 Performance characteristics

The achievable performance of the method is given in [Annex A](#) showing the results obtained by a European comparison study carried out for a sample of wood chips and a sample of an exhausted olive residue. These two samples represent the extremity of the method. The wood chip sample represents samples with low contents of water soluble chloride, sodium and potassium and the olive residue samples with high amounts of water soluble chloride and potassium.

11 Test report

The test report shall include at least the following information:

- a) a reference to this International Standard (ISO 16995);
- b) the identification of the laboratory performing the test and the date of the test;
- c) the identification of product (or sample) tested;
- d) the detection method(s) used for the determination;
- e) the results of the test including the basis in which they are expressed, as indicated in [Clause 9](#);
- f) any unusual features noted during the determination;
- g) any operation not included in this International Standard, or regarded as optional.

Annex A (informative)

Performance data

A comparison study was carried out by laboratories in Austria, Belgium, Denmark, Finland, Germany, Ireland, Italy, The Netherlands, Spain, Sweden and the United Kingdom. The variety of instruments and other analytical conditions were used in accordance with the quality parameters specified in the method.

The tests were carried out using two samples, wood chips and exhausted olive residues produced in the EU-project BioNorm according to prEN 14780:2006. The “wood chips” sample was made of German coniferous wood chips; the chips were dried and milled to 1 mm by means of cutting mill. The “exhausted olive residues” sample was obtained from olive oil industry in Spain from a typical outdoor storage facility. In the original sample, stones and other natural impurities were present. These impurities and stones were removed manually and the sample was prepared from the residues in two steps using a coarse cutting mill equipped with a 10 mm sieve and a laboratory cutting mill equipped with tungsten carbide (WC) cutting tools and a 1 mm sieve.

All data are reported on dry basis.

The performance data according to ISO 5725-2 are presented in [Tables A.1](#), [A.2](#) and [A.3](#).

NOTE A guideline can be found in ISO 16993:2015, Annex C on how to use these validation parameters.

Table A.1 — Performance data for water soluble chloride

Sample	n	l	O %	$\bar{x}$ mg/kg	s_R mg/kg	$C_{V,R}$ %	s_r mg/kg	$C_{V,r}$ %
Wood chips	7	35	15	26	13	45	5	19
Exhausted olive residues	7	35	0	2 100	150	7,4	49	2,4
Definitions of the symbols n is the number of laboratories after outlier elimination l is the number of outlier free individual analytical values O is the percentage of outlying values from replicate determination $\bar{x}$ is the overall mean s_R is the reproducibility standard deviation $C_{V,R}$ is the coefficient of the variation of the reproducibility s_r is the repeatability standard deviation $C_{V,r}$ is the coefficient of the variation of the repeatability								

Table A.2 — Performance data for water soluble sodium

Sample	n	l	O %	$\bar{x}$ mg/kg	s_R mg/kg	$C_{V,R}$ %	s_r mg/kg	$C_{V,r}$ %
Wood chips	There is not enough data available							
Exhausted olive residues	8	40	0	99	16	16	8	8,4
NOTE See Table A.1 for definitions of the symbols.								